

Research on the effect of hydrogen charging on the extreme tip bending performance of ultrahigh strength automotive steel plate

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In this study, the effect of hydrogen charging treatment on the extreme tip bending performance of ultra-high strength automotive steel plates was investigated. After electrochemical hydrogenation, the extreme tip bending angle of the steel plate decreased to varying degrees, and the downward trend began to converge after 6 hours of hydrogenation. After 6 hours of hydrogen charging, the extreme tip bending angle of 22MnB5 bare plate decreased by 49.2%, 22MnB5-NbV by 36.7%, 34MnB5 by 49%, 34MnB5-V by 34.6%, and 34MnB5-Nb by 30.8% respectively. In addition, Al-Si coated plates need to be pre-bended treatment before the experiment to produce micro cracks on the surface to get the chance of the hydrogen getting into the plate. The influence of hydrogen overflow on the extreme tip bending angle was also studied, and it was found that the filled hydrogen will overflow completely after 24 hours. Therefore, this report summarizes the effect of hydrogen content on the extreme tip bending angle, hoping to provide positive guidance for future research on the hydrogen embrittlement resistance of high-strength steels.

Keywords: Hydrogen charging; Extreme tip bending; Hydrogen embrittlement.

1. Introduction

With the upgrading of the global automotive industry and the comprehensive rise of manufacturing and marketing of China's new energy vehicles, the steels used in automobiles are gradually shifting towards high strength and elongation. The application volume of high-strength steels (USS), advanced high-strength steels (AHSS), or ultra-high-strength steels (UHSS) have been becoming increasingly widespread. It is well known that hydrogen embrittlement behavior occurs when the strength of steels exceeds 1000MPa, which has always affected the research and development of UHSS of 1500MPa and above^[1-4]. How to enhance the hydrogen embrittlement resistance of HSS, and objectively

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and rapidly evaluate its effect degree, occurring mechanism and accident time, are the hottest topics of considerable concerns of the vehicle safety among domestic and international scholars ^[5-9]. The extreme tip bending test is an important parameter for quickly evaluating toughness and the minimum bending angle at which automotive steel plates will fracture, which is crucial for the automotive industry. If the extreme tip bending performance can be used to evaluate the hydrogen embrittlement resistance, it will greatly improve the efficiency and accuracy of evaluating the hydrogen embrittlement resistance. Therefore, the author conducted hydrogen charging at different current and time, and studied the effect of hydrogen charging treatment on the extreme tip bending performance of UHSS from a macroscopic perspective.

2. Experimental Materials and Methods

2.1. Experimental Steels and Treatment

The UHSS used were all PHS (Press Hardened Steel), including 2 types of uncoated 22MnB5 steel plates, 3 types of uncoated 34MnB5 steel plates, and one type of 22MnB5-AlSi steel plate. The thickness of both 22MnB5 and 34MnB5 steel plates was 1.5 mm, and the thickness of 22MnB5-AlSi steel plate was 1.0 mm. The chemical composition is shown in Table 1.

Steel	С	Si	Mn	Al	Cr	В	Ti	Nb	V
22MnB5-1	0.22	0.24	1.30	0.03	0.13	0.002	0.031	0	0
22MnB5-2	0.23	0.23	1.32	0.03	0.14	0.002	0.033	0.04	0.04
34MnB5-1	0.33	0.31	1.45	0.03	0.55	0.002	0.030	0	0
34MnB5-2	0.34	0.31	1.50	0.03	0.59	0.002	0.035	0	0.20
34MnB5-3	0.34	0.64	1.28	0.03	0.60	0.002	0.028	0.08	0
22MnB5-AlSi	0.22	0.24	1.30	0.03	0.13	0.002	0.031	0	0

Table 1 The chemical compositions of the studied steels (in wt%)

The steel plates were heated at 930 °C, and held for 5 minutes for austenitization ^[10]. Then quenched by mold pressing to room temperature, and the microstructure after quenched was all martensite. The tensile strength of 22MnB5 is about 1500MPa, and the tensile strength of 34MnB5 is about 1900MPa. Cut the quenched plate into standard specimen size which was 60 mm at rolling direction and 30 mm at perpendicular rolling direction. Then, polish its surface in sequence with 180-1200 mesh SiC sandpaper until the surface is smooth and free of oxide skin.

The polished samples of uncoated 22MnB5 and 34MnB5 steel plates were subjected to hydrogen charging treatment. The hydrogen charging solution was NaOH with a concentration of 1mol/L, and 1g/L of CH4N2S alkaline aqueous solution was added. The hydrogen charging current was 20mA/cm², and the hydrogen charging time was 0.5-24 hours. The steel type to be tested was the cathode, and the carbon rod was the anode. In

addition, the 22MnB5-AlSi steel plate needs to be pre bent before hydrogen charging treatment.

After hydrogen charging of the sample steel, the extreme tip bending test shall be carried out immediately, and the test execution standard is T/CSAE 154-2020 "Ultra high strength automobile steel plate-Extreme tip bending test".

This paper introduces the sensitivity coefficient I_d of extreme tip bending, and its calculation formula is shown in Equation (1),

$$I_d = 100\% \times (D_s - D_f) / D_s$$
(1)

Where D_s is the extreme tip bending without hydrogen charging, and D_f is the extreme tip bending after 6 hours of hydrogen charging. The smaller the sensitivity coefficient I_d of extreme tip bending, the better the hydrogen embrittlement resistance.

2.2. Experimental equipment

The hydrogen charging power supply is a domestically produced equipment, model UNI-T-UPT1310, with an output current of 20mA/cm².

The testing equipment for extreme tip bending performance is a universal tensile testing machine with an extreme tip bending fixture, and the testing principle diagram is shown in Figure 1. The brand and model of the universal stretching machine is Zwick/Roell-Z100, with a maximum range of 100kN and a set pressing speed of 15mm/min.

The thermal desorption spectroscopy (TDS) equipment consists of a vacuum pumping system, glass tube, heating furnace, and quadrupole mass spectrometer (INFICON, USA). During testing, the vacuum pressure is below 1.2×10^{-4} Pa, with a heating rate of approximately 10K/min, and a maximum testing temperature of 800°C. A quadrupole mass spectrometer is used to measure the hydrogen content absorbed from the sample during the desorption process. Measure the hydrogen content absorbed from the sample during the temperature ramp-up process using a quadrupole mass spectrometer.



Figure 1. The principle of extreme tip bending test

In Figure 1, F is the bending load, S is the punch displacement, D is the roller diameter, L is the roller distance, I is the sample length, b is the sample width.

2.3. Experimental Plan

Three experimental plans were designed as following:

Plan 1, Hydrogen charging \rightarrow Extreme tip bending test. 22MnB5-1, 22MnB5-2, 34MnB5-1, 34MnB5-2 were hydrogen charged for 0.5h, 1h, 3h, 6h, 12h, and 24h, respectively, with a hydrogen charging current of 20mA/cm2. Immediately after hydrogen charging, conducted the extreme tip bending test to obtain the relationship between hydrogen charging time and the extreme tip bending angle.

Plan 2, Pre bending \rightarrow Hydrogen charging \rightarrow Extreme tip bending test. The 22MnB5-AlSi steel plate was pre bent at 20° to produce micro cracks on the surface, and then subjected to hydrogen charging treatment for 5min, 10min, 20min, 60min, and 120min, with a hydrogen charging current of 10mA/cm2. Immediately after hydrogen charging, conducted the extreme tip bending test to obtain the relationship between hydrogen charging time and the extreme tip bending angle.

Plan 3, Pre bending \rightarrow Hydrogen charging \rightarrow Placement \rightarrow Extreme tip bending test. The 22MnB5-AlSi steel plate was pre bent at 20° to produce micro cracks on the surface, and then subjected to 20 minutes of hydrogen charging treatment with a hydrogen charging current of 10mA/cm2. Then, it was placed for 3h, 6h, 15h, 24h, 48h, 72h, and 96h respectively. The specimen was further subjected to extreme tip bending test to obtain the relationship between the placement time and the extreme tip bending angle.

3. Result and discussion

3.1. Hydrogen charging \rightarrow Extreme tip bending test

According to the plan 1, extreme tip bending test were conducted after hydrogen charging, with a hydrogen charging current of 20mA/cm². The results are shown in Table 2 and Figure 2 and 3. According to the results, the longer the hydrogen charging time, the smaller the extreme tip bending angle gradually decreases, and the extreme tip bending angle tends to stabilize after 6 hours of hydrogen charging. Due to the effective enhancement of hydrogen embrittlement resistance by elements Nb and V ^[10-14], the extreme tip bending angle of 22MnB5-2 after hydrogen charged was higher than the 22MnB5-1, and the 34MnB5-2 and 34MnB5-3 were higher than the 34MnB5-1. After 6 hours of hydrogen charging, the extreme tip bending sensitivity coefficients of 22MnB5-1 and 22MnB5-2 were 49.2% and 36.7%, respectively, and the extreme tip bending sensitivity coefficients of 1900-1, 1900-2, and 1900-3 were 49.0%, 34.6%, and 30.8%, respectively. The smaller the extreme tip bending sensitivity coefficient Id, the better the hydrogen embrittlement resistance. Therefore, the I_d value of PHS with excellent resistance to hydrogen embrittlement is lower than that of ordinary PHS.

In order to evaluate the hydrogenation effect, samples of 34MnB5-1 were selected and subjected to TDS test to measure the hydrogen content after 1 hour and 6 hours of hydrogenation, as shown in Figure 4 and Figure 5. The results indicate that the hydrogen content is approximately 11 ppm after 1 hour of hydrogenation, and approximately 32 ppm after 6 hours.

Table 2 Results of extreme up bending test after hydrogen charging							
Charging time	Extreme tip bending angle (deg)						
(h)	22MnB5-1	22MnB5-2	34MnB5-1	34MnB5-2	34MnB5-3		
No charging	59	60	51	52	52		
0.5	48	50	42	48	50		
1	44	46	38	43	44		
3	36	40	32	38	40		
6	30	38	26	34	36		
12	29	37	25	32	35		
24	29	37	25	32	35		
I_d	49.2%	36.7%	49.0%	34.6%	30.8%		



Figure 2 The relationship between 22MnB5 hydrogen charging time and extreme tip bending angle



Figure 3 The relationship between 34MnB5 hydrogen charging time and extreme tip bending angle



Sample Weight: 20.49g 0.012 104°C (s/md) 0.000 0.000 0.004 0.004 0.004 0.398ppm 548°C 0.000 100 200 300 400 600 ò 500 700 Temperature/°C

Figure 4 TDS test results of 34MnB5-1 sample after hydrogen charging for 1 hour

Figure 5 TDS test results of 34MnB5-1 sample after hydrogen charging for 6 hours

3.2. Pre bending \rightarrow Hydrogen charging \rightarrow Extreme tip bending test

According to Plan 2, a pre-bending of 20° was performed on the 22MnB5-AlSi sheet to induce microcracks on the surface of the material, as shown in Figure 6. The pre bending sample were subjected to the extreme tip bending test after hydrogen charging, with a hydrogen charging current of 10mA/cm². The results of the test are shown in Figure 7.





3.3. Pre bending \rightarrow Hydrogen charging \rightarrow Placement \rightarrow Extreme tip bending test

According to Plan 3, a pre-bending of 20° was performed on the 22MnB5-AlSi plate to induce microcracks on the surface of the material. The sample was then hydrogen charged for 20 minutes until it reached a state of full or over-saturation with hydrogen, before being left for up to a maximum of 96 hours prior to conducting the extreme tip bending test. The results of the test are shown in Figure 8. From the results, it can be observed that within 6 hours of hydrogenation, there is no significant change in the angle of the extreme tip

bending, indicating an over-saturation state of hydrogenation previously. Furthermore, after 24 hours of placement, the angle tends to stabilize and return to the state without hydrogenation, indicating that all hydrogen has diffused out after 24 hours.



Figure 8 The relationship between the placement time of 22MnB5-AlSi plate after hydrogen charging and the extreme tip bending angle

4. Conclusion

(1) As the hydrogen charging time increases, the extreme tip bending angle gradually decreased The uncoated 22MnB5 and 34MnB5 steel plates would converge after 6 hours of hydrogen charging at a hydrogen charging current of 20mA/cm^2 , indicating that hydrogen charging was beginning to saturate. After 6 hours of hydrogen charging, the extreme tip bending sensitivity coefficients (I_d) of 22MnB5-1 and 22MnB5-2 were 49.2% and 36.7%, respectively. The extreme tip bending sensitivity coefficients of 1900-1, 1900-2, and 1900-3 were 49.0%, 34.6%, and 30.8%, respectively. It can be seen that the I_d value of PHS with excellent resistance to hydrogen embrittlement is lower than that of ordinary PHS. Therefore, using this method can easily and quickly evaluate the hydrogen embrittlement resistance of ultra-high strength steel plates.

(2) After the appearance of microcracks on the surface of 22MnB5-AlSi steel plate, with the hydrogen charging current of 10mA/cm², under 20 minutes of hydrogen charging, the extreme tip bending angle will be converged, indicating that the surface would accelerate saturation after microcracks.

(3) After 24 hours of placement following hydrogenation, the angle of the extreme tip bending returned to the value of the non-hydrogenated state, indicating that all hydrogen introduced during hydrogenation had been completely eliminated.

The significance of this paper is to study the effect of hydrogen on the performance of ultra-high strength steel plates by changing the extreme tip bending angle through hydrogen charging and hydrogen emission, providing new ideas for evaluating the toughness and hydrogen embrittlement resistance of ultra-high strength steel plates.

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